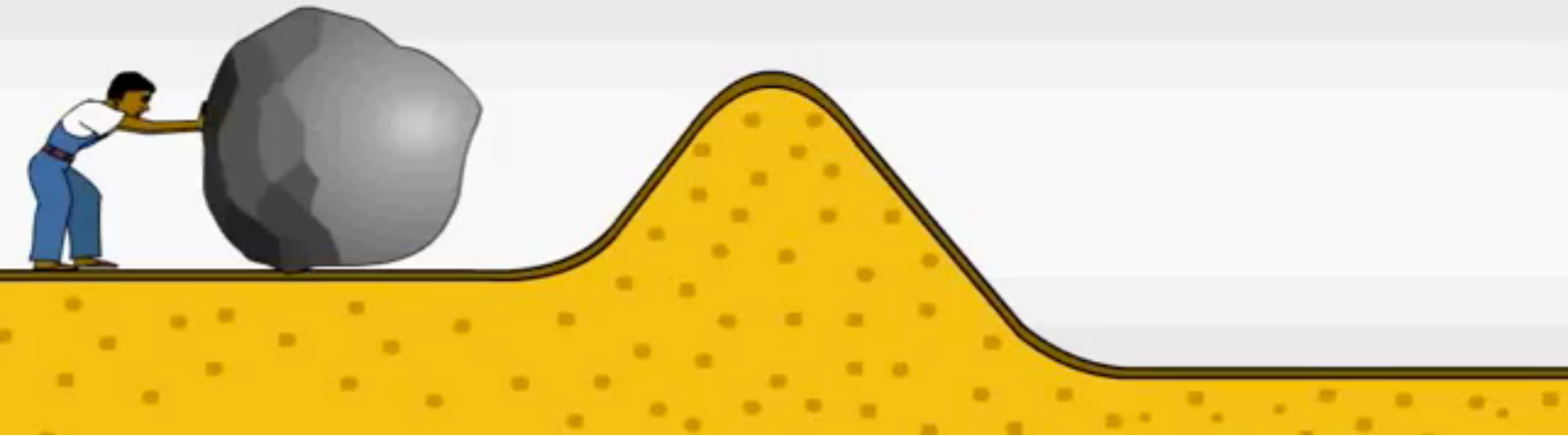
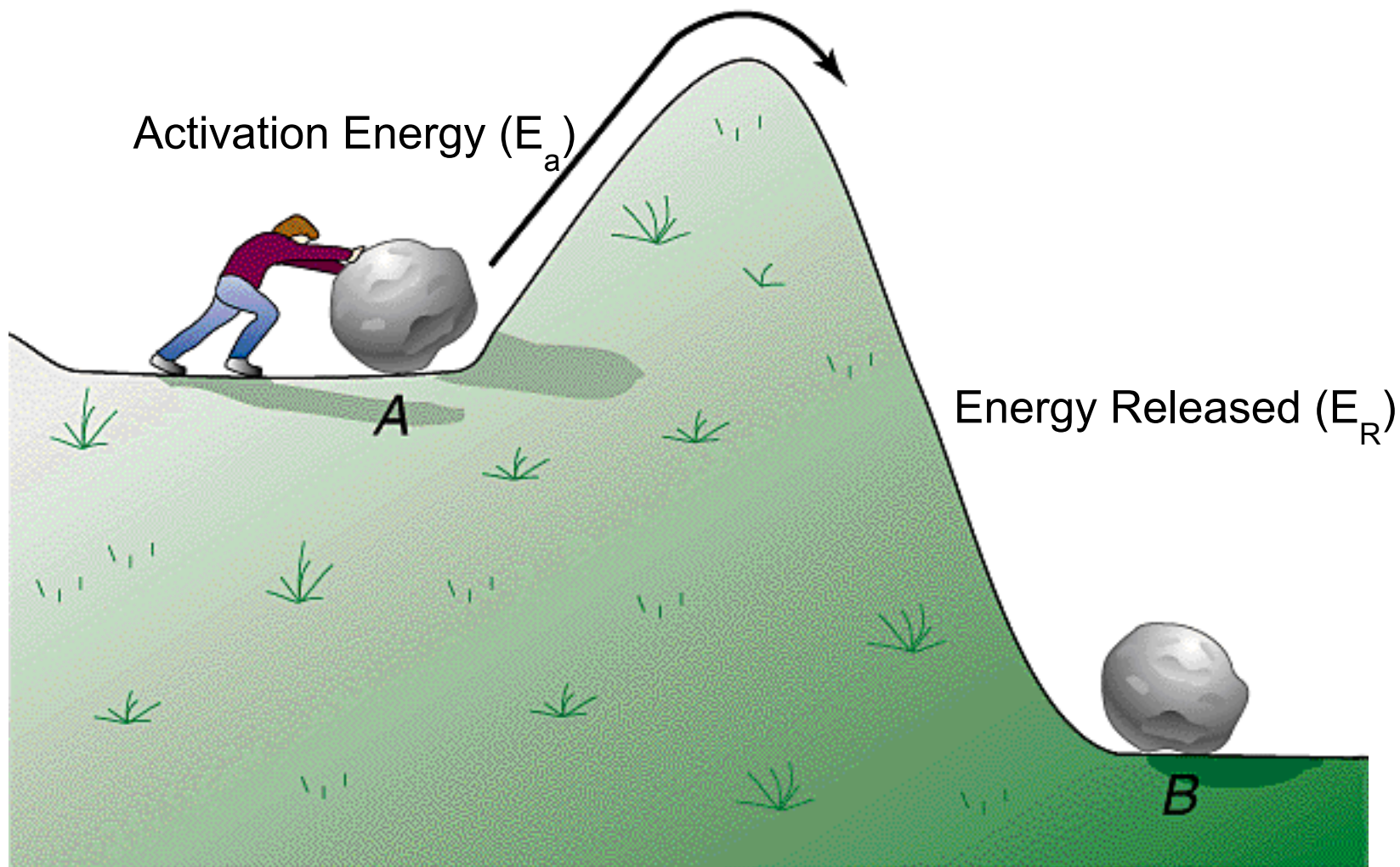


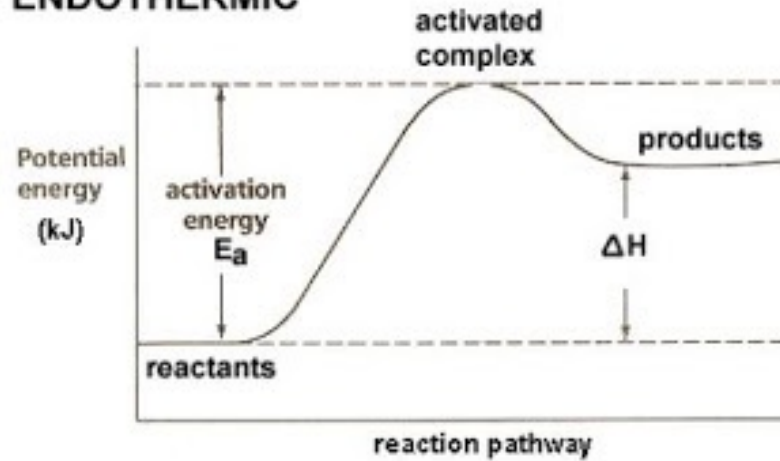
# Activation Energy and Enzymes



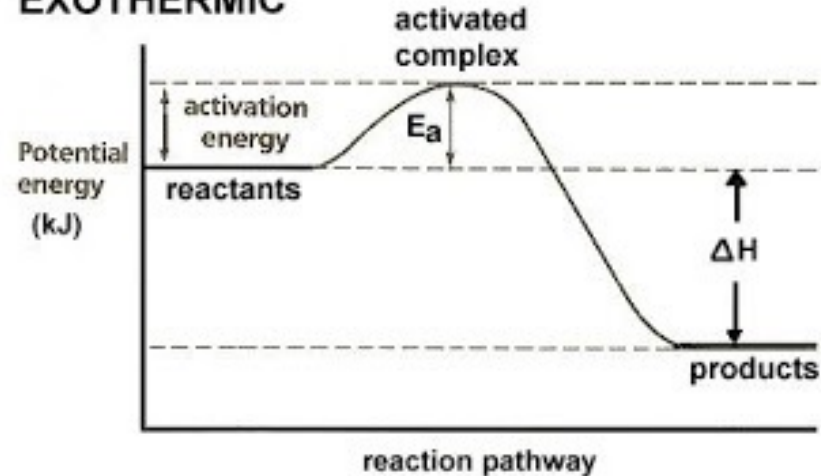


# Activation Energy ( $E_a$ )

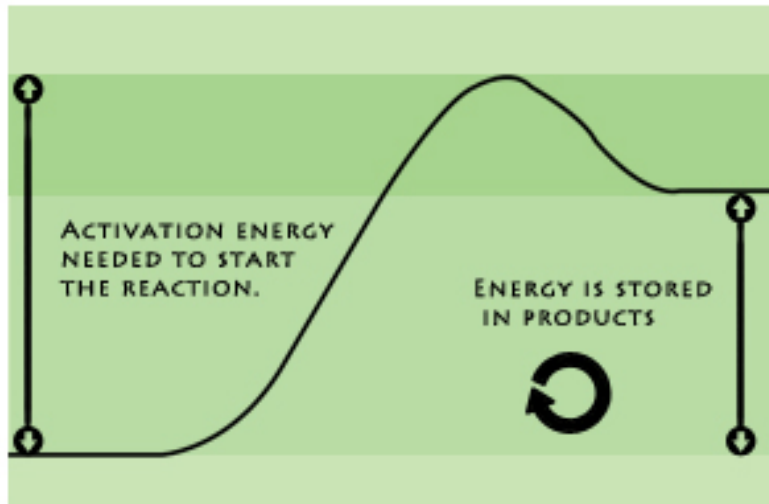
**ENDOTHERMIC**



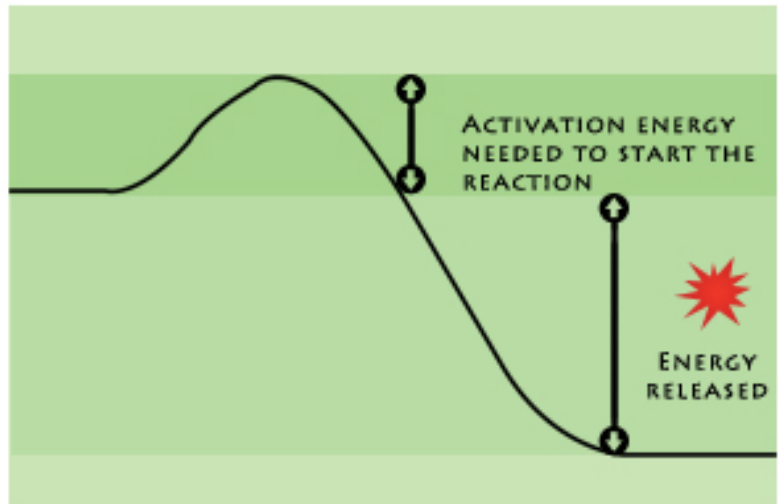
**EXOTHERMIC**



# ENDERGONIC



# EXERGONIC



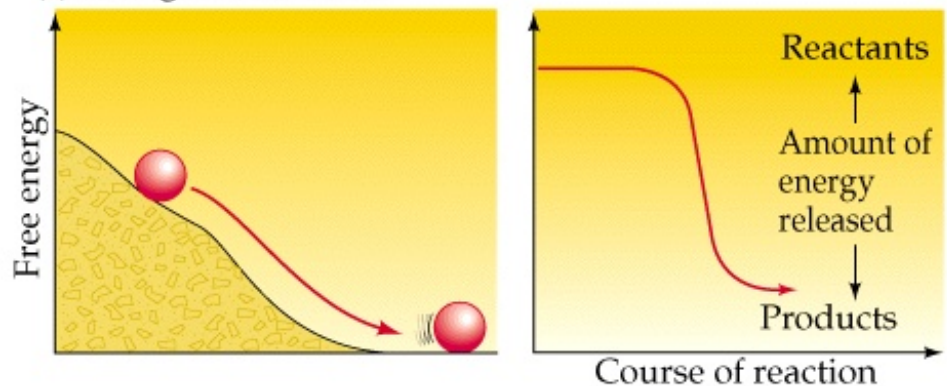
# Exergonic vs Endergonic

In reactions

If products release more energy than the input of energy  $\rightarrow$  **Exergonic**

(energy to break bonds  $<$  energy released when new bonds form)

(a) Exergonic reaction

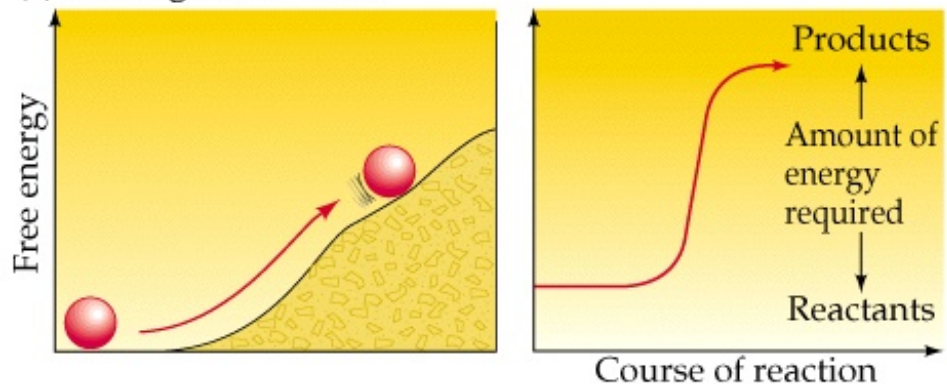


In reactions

If products release less energy than the input of energy  $\rightarrow$  **Endergonic**

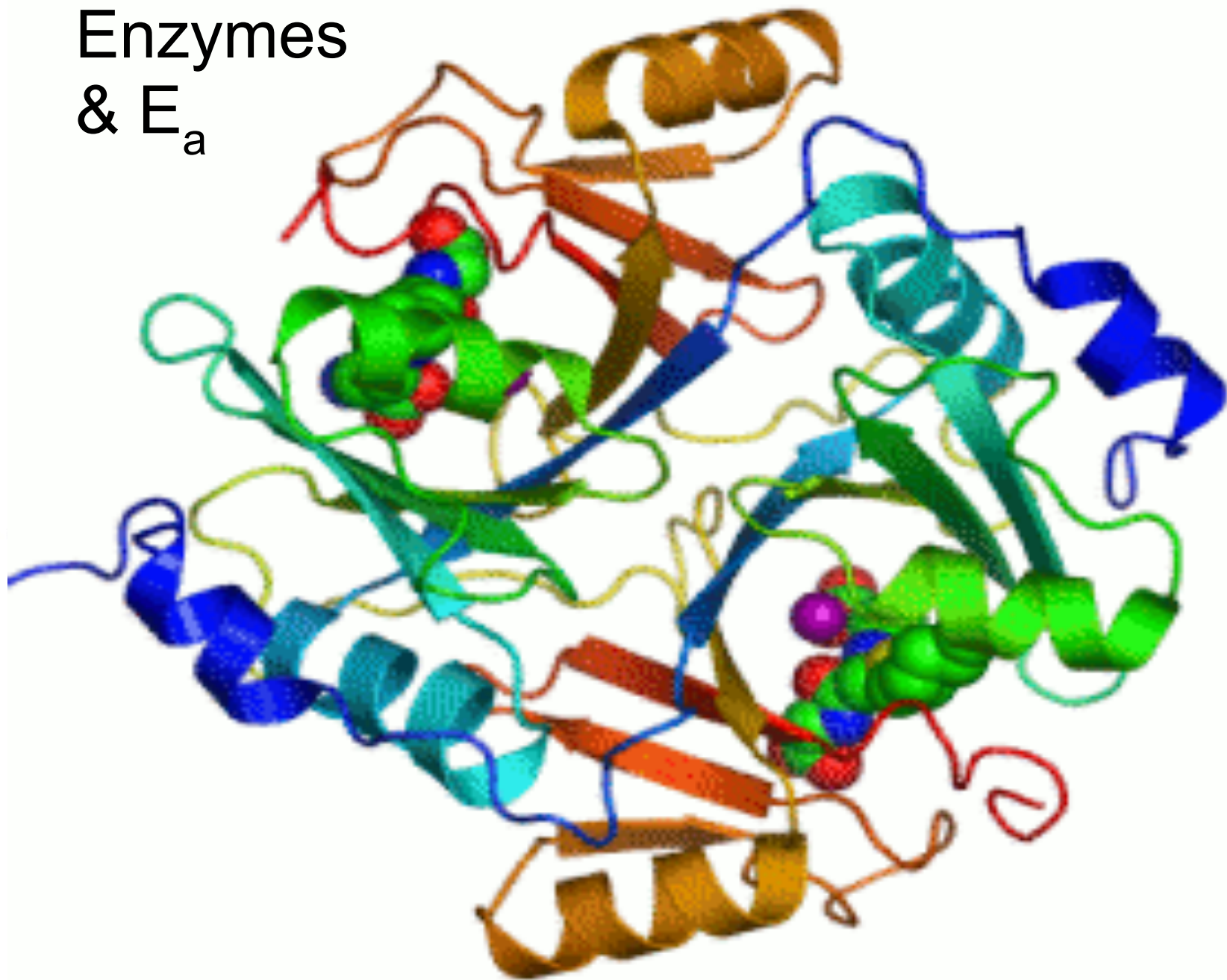
(energy to break bonds  $>$  energy released when new bonds form)

(b) Endergonic reaction

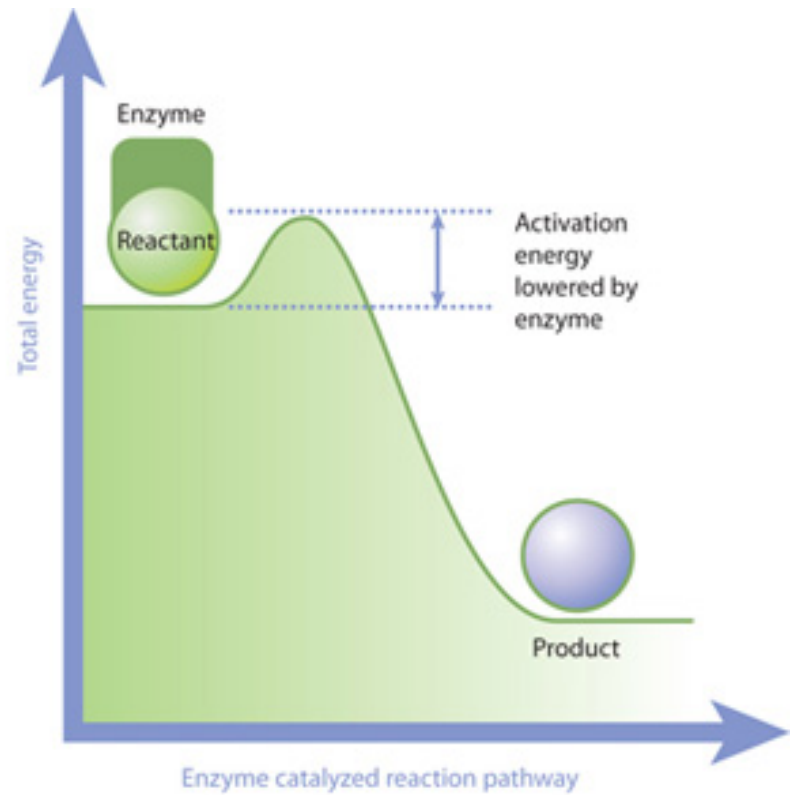
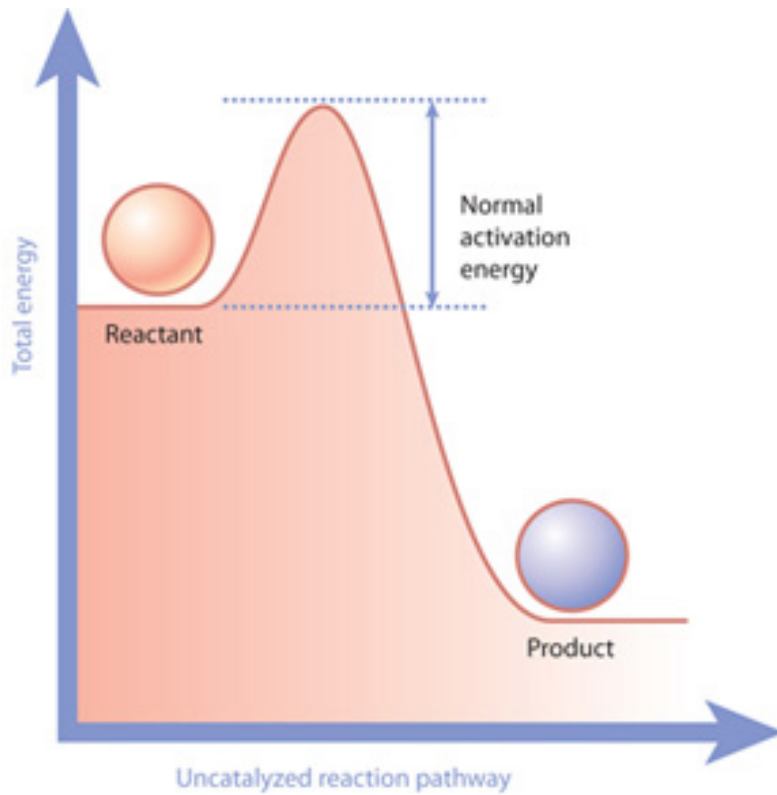




# Enzymes & $E_a$



# How is $E_a$ Overcome?

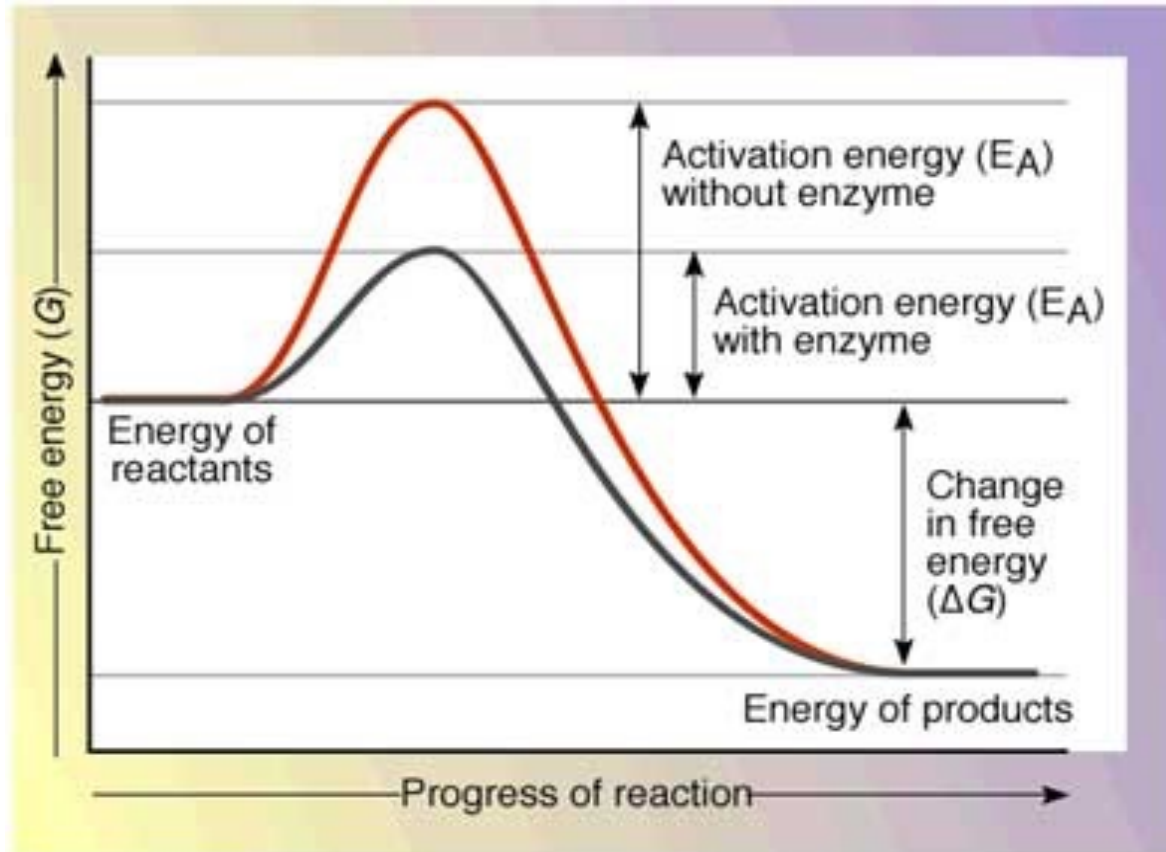


# How is $E_a$ Overcome?

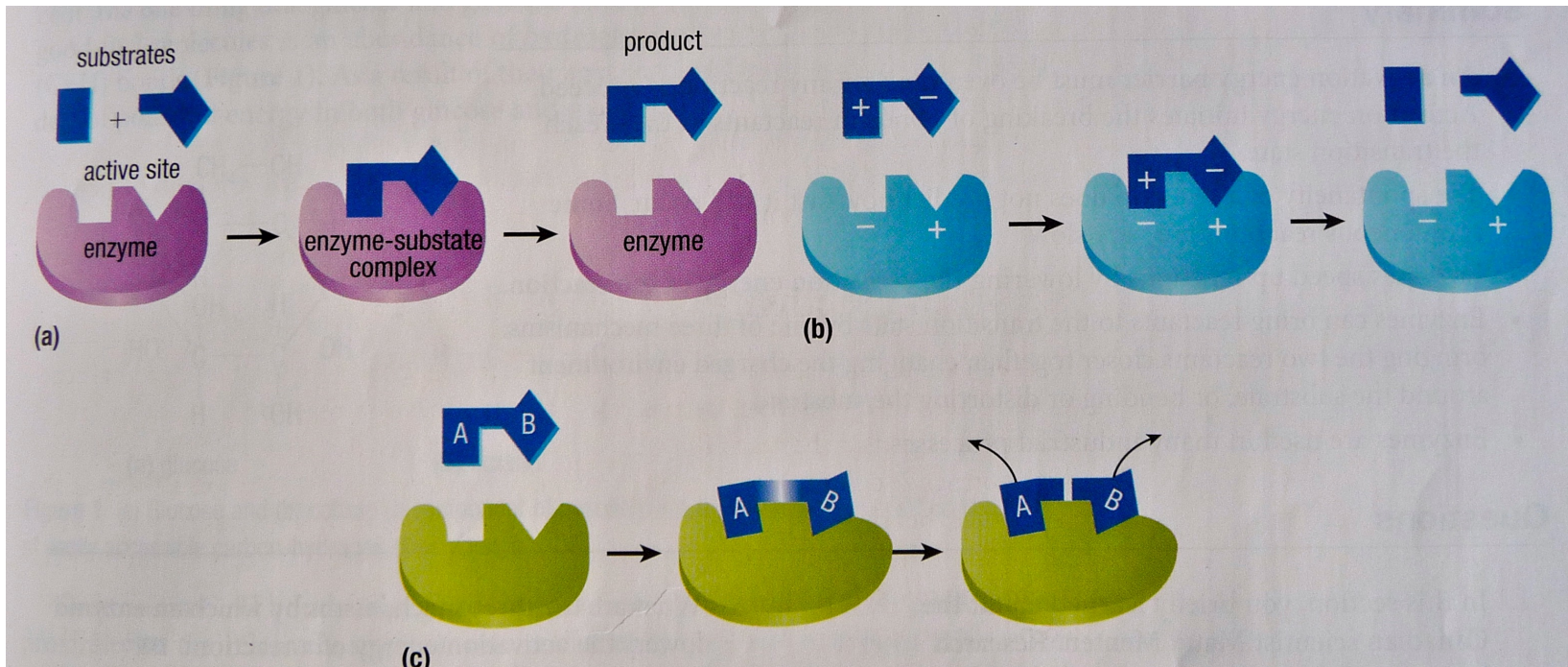
- enzymes lower the activation energy
- increasing the likelihood of a reaction occurring
- since enzymes are substrate specific, reactions are controlled.



# Enzymes and $E_a$



# How do Enzymes lower $E_a$ .



# How do Enzymes lower $E_a$ .

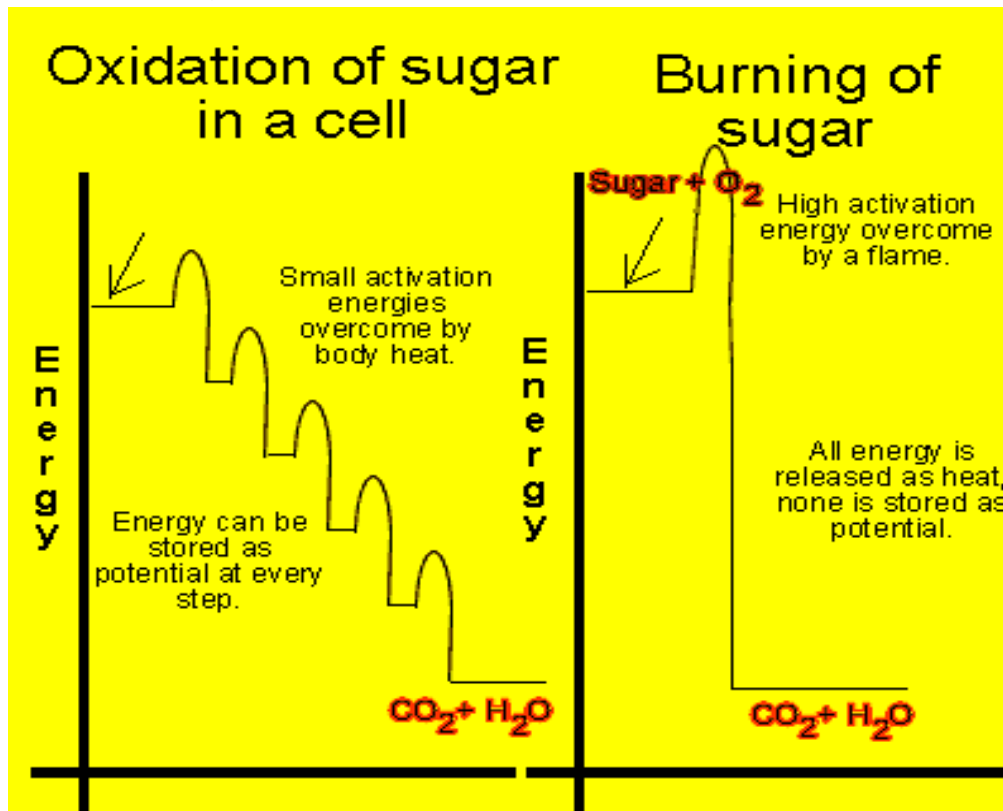
- » enzymes bring substrate in close proximity with proper orientation
- » enzymes create an environment to favourable for substrates to react
- » enzymes place strains on existing bonds

# Metabolic Pathways and Enzymes revisited



# Metabolic Pathways

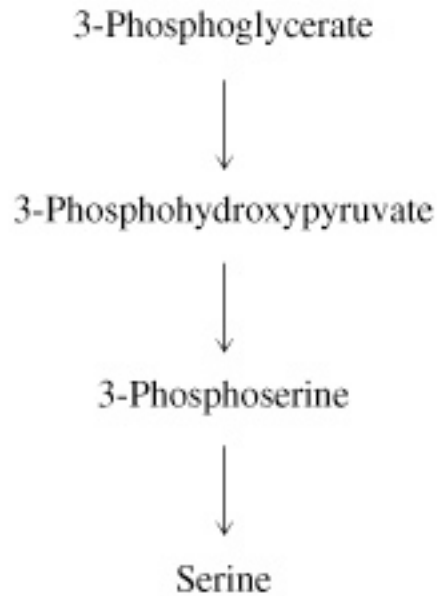
- » 'Metabolism' termed by Schwann referring to chemical changes in cells
- » Chemical changes occur in incremental steps that are enzyme driven



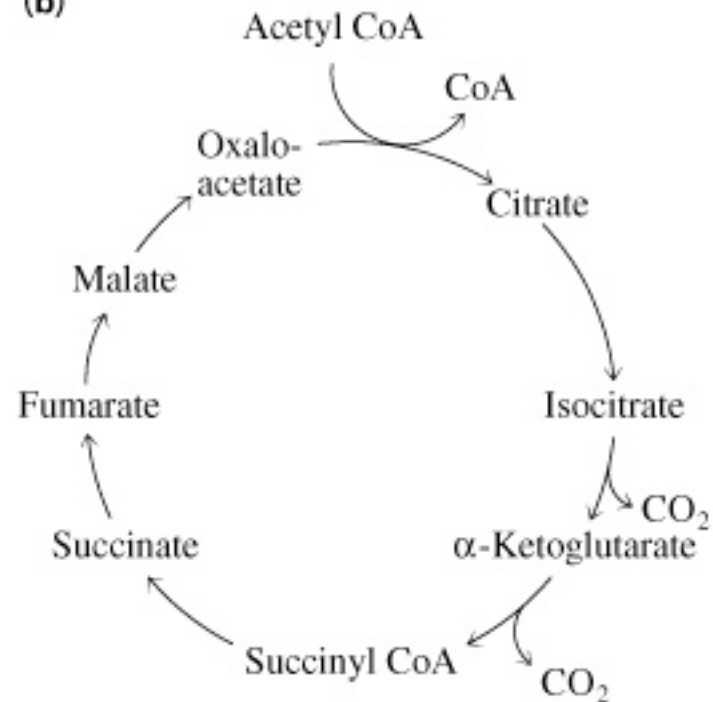
# Pathways

» Some are linear while other can be cycles

(a)



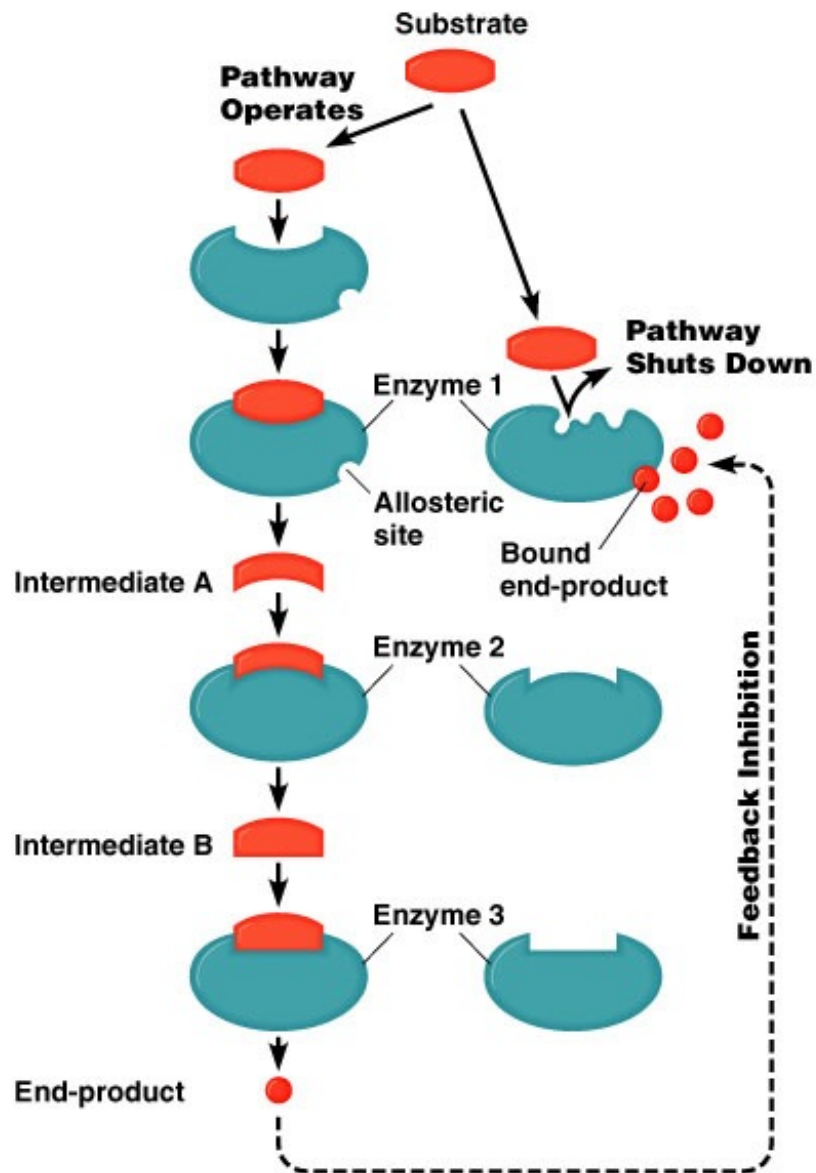
(b)



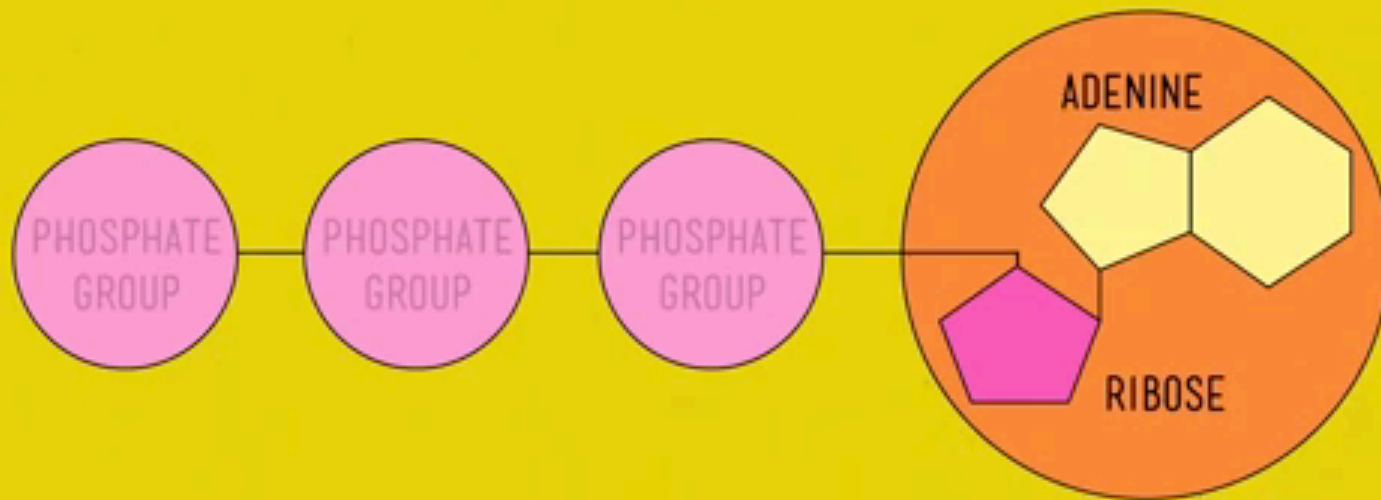


# Pathways are regulated Feedback Inhibition

- often used to regulate multi-step metabolic pathways
- A product of a pathway acts as an allosteric inhibitor of the first enzyme of the pathway



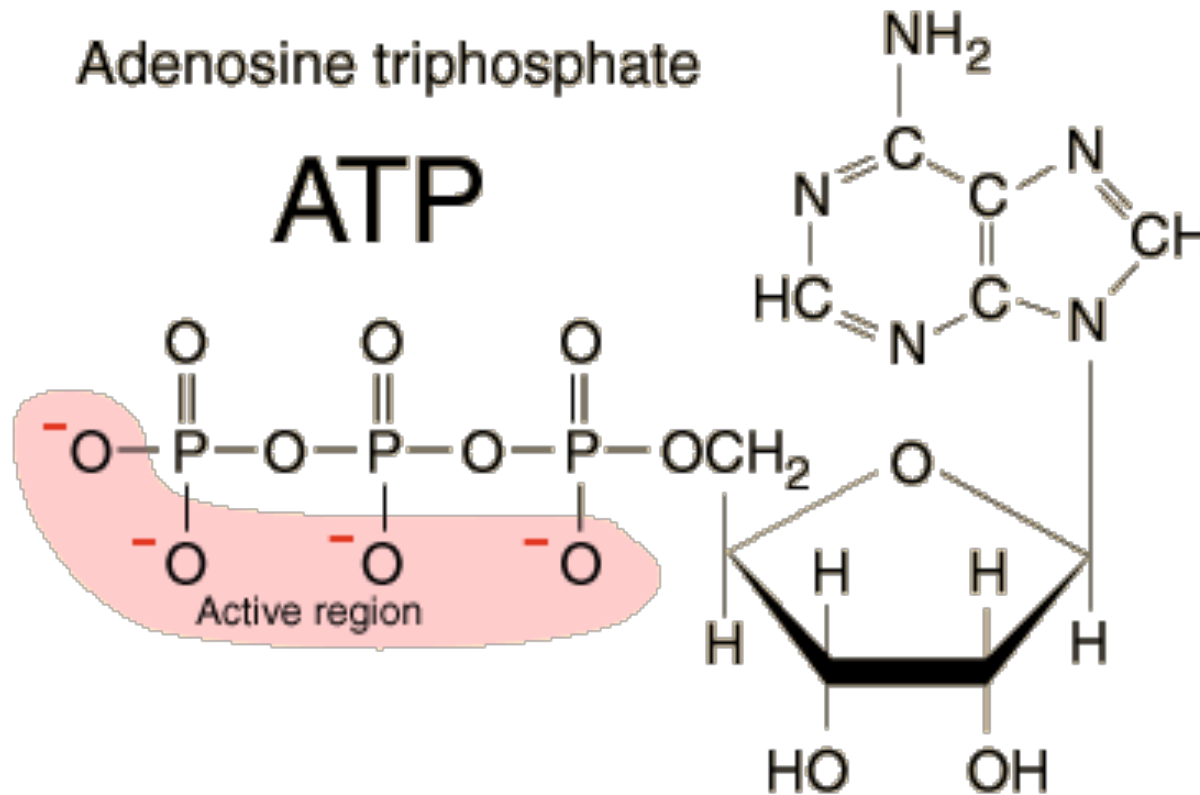
# ATP (ADENOSINE TRIPHOSPHATE)



# Structure of ATP

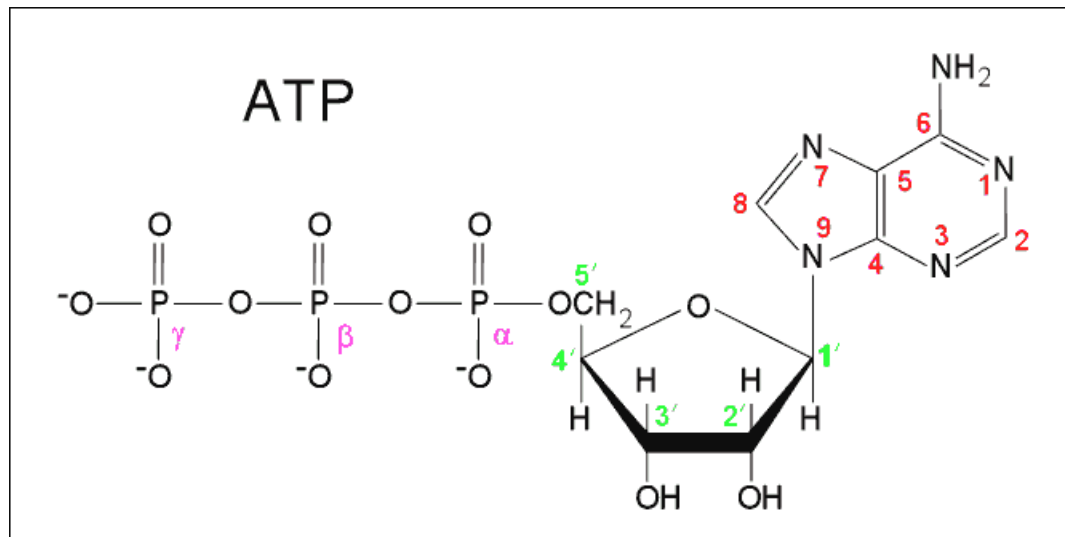
Adenosine triphosphate

## ATP



# Adenosine Triphosphate

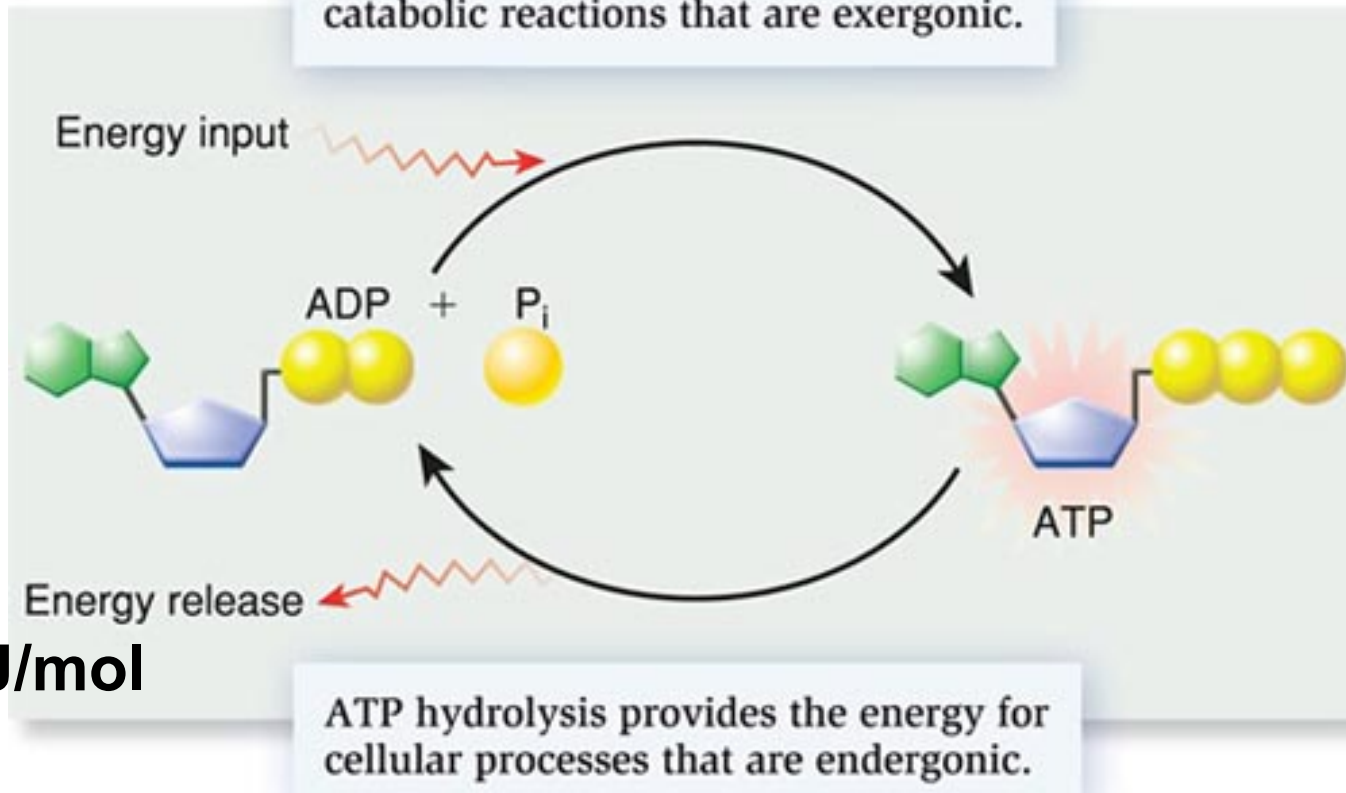
- All cells in every living organism use the same energy carrier (ATP) for almost all Energy-requiring actions.



# ATP cycle

**Phosphorylation**

The energy to make ATP comes from catabolic reactions that are exergonic.

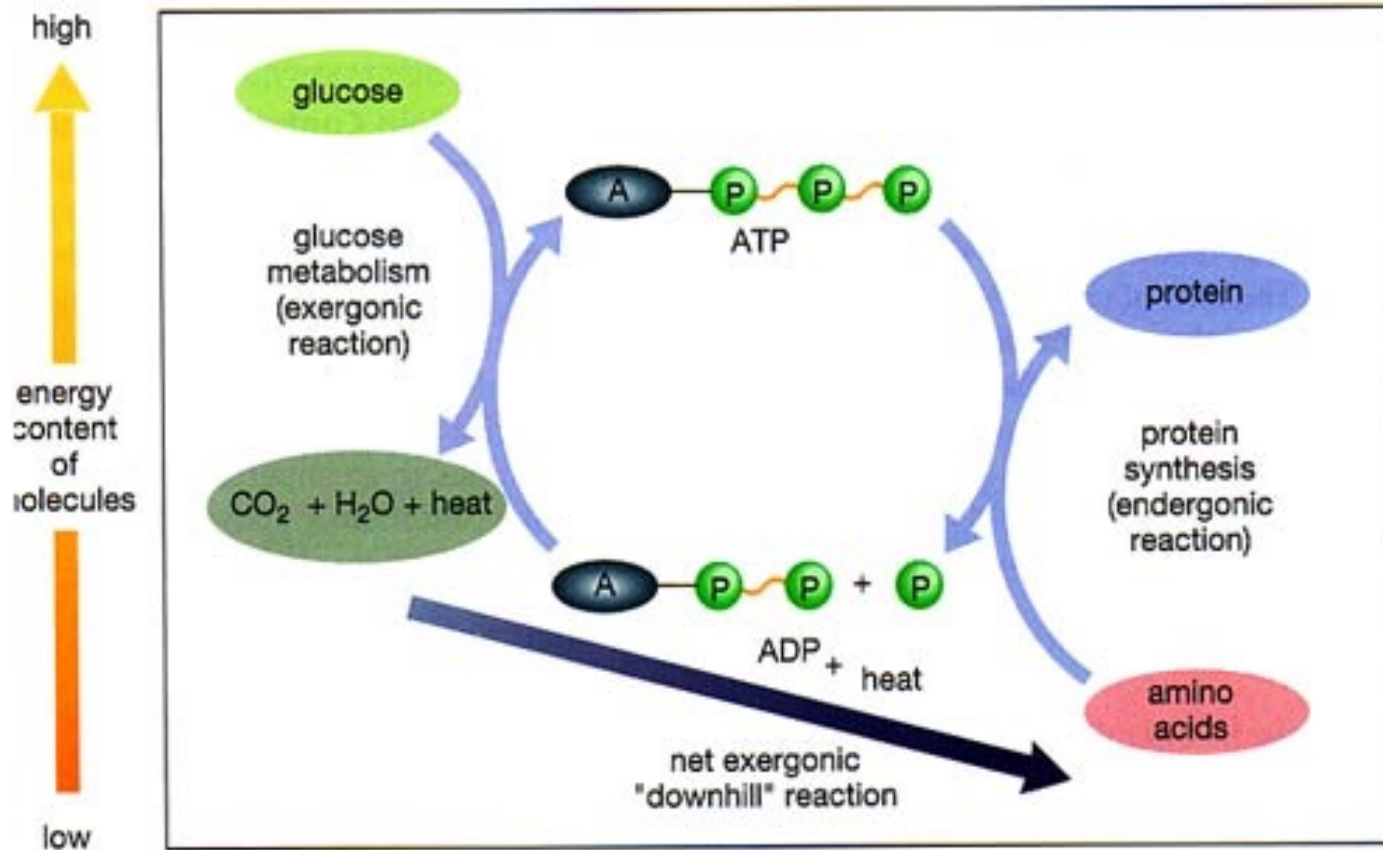


$$\Delta E = 30 \text{ kJ/mol}$$

**Hydrolysis**



# Energy Coupling



For reactions to work in cells, an exergonic reaction is required to release energy to power an endergonic reaction. This is **COUPLING**.

# Why do cells use ATP?

- Why not use other energy-rich molecules directly?
  - manageable amount of Energy
  - “universal” - is able to couple with many different endergonic reactions
  - usable in single reaction (unlike heat that will affect multiple reactions at once)