

- <u>Lipids</u> macromolecules that do not dissolve in water, such as fats, oils, steroids
- used for:
- » long-term nutrient storage
- » insulation
- » cushioning of internal organs
- » hormones to send messages around body
- » important part of cell membrane (mb)

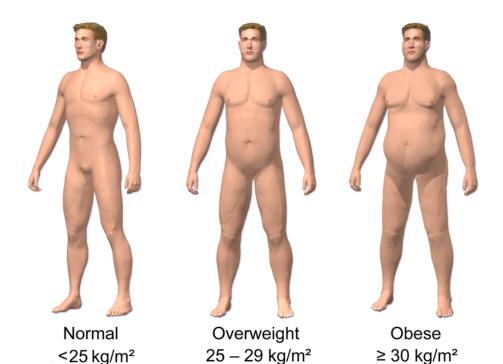
1g fat =  $2.25 \times energy$  as 1g carbohydrate

- · more efficient at storing energy
- carbohydrate storage in cells is 6x more massive than lipids of the same energy (ie. add H2O) - birds use this as energy storage for flight
- poor conductor of heat, —> insulates

# Body Mass Index

https://www.cdc.gov/healthyweight/assessing/bmi/adult\_bmi/english\_bmi\_calculator/bmi\_calculator.html

$$ext{BMI} = rac{ ext{mass}_{ ext{kg}}}{ ext{height}_{ ext{m}}^2} = rac{ ext{mass}_{ ext{lb}}}{ ext{height}_{ ext{in}}^2} imes 703$$



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# Body Mass Index

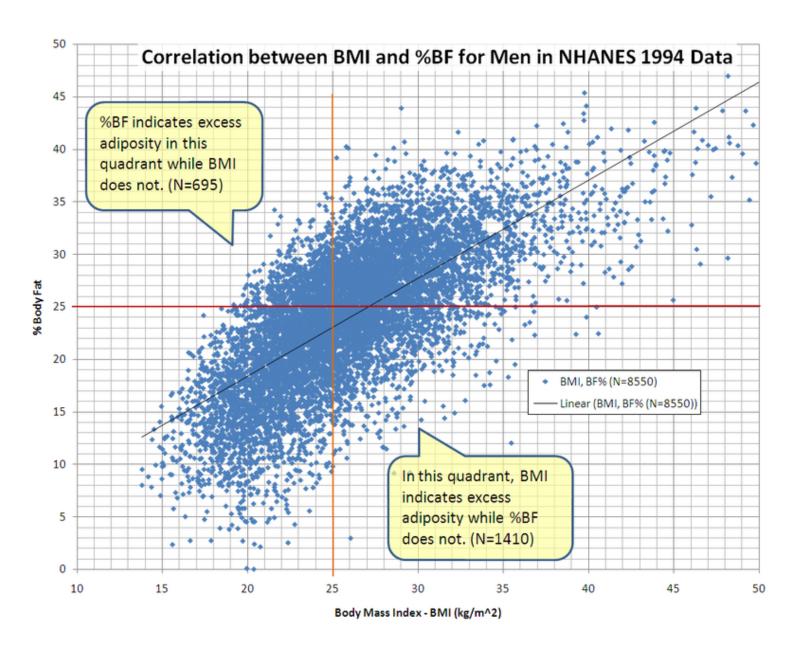
Category	BMI (kg/m²)	
	from	to
Very severely underweight		15
Severely underweight	15	16
Underweight	16	18.5
Normal (healthy weight)	18.5	25
Overweight	25	30
Obese Class I (Moderately obese)	30	35
Obese Class II (Severely obese)	35	40

# Body Mass Index

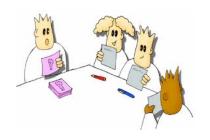
#### Health risks from having a high BMI

- High blood pressure (Hypertension)
- Type 2 diabetes
- Coronary heart disease
- Stroke
- Gallbladder disease
- Osteoarthritis (a breakdown of cartilage and bone within a joint)
- Sleep apnea and breathing problems
- Chronic inflammation
- Some cancers (endometrial, breast, colon, kidney, gallbladder, and liver)
- Low quality of life
- Mental illness such as clinical depression, anxiety, and other mental disorders
- Body pain and difficulty with physical functioning

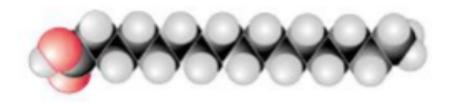
#### **Limitations to BMI**

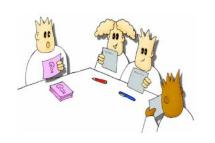


» <a href="https://www.muscleandstrength.com/">https://www.muscleandstrength.com/</a> tools/measure-bodyfat



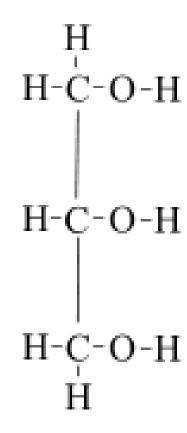
- 1 Build a fat acid chain 8 carbons long.
- a) Describe the molecule's overall shape.

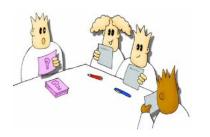




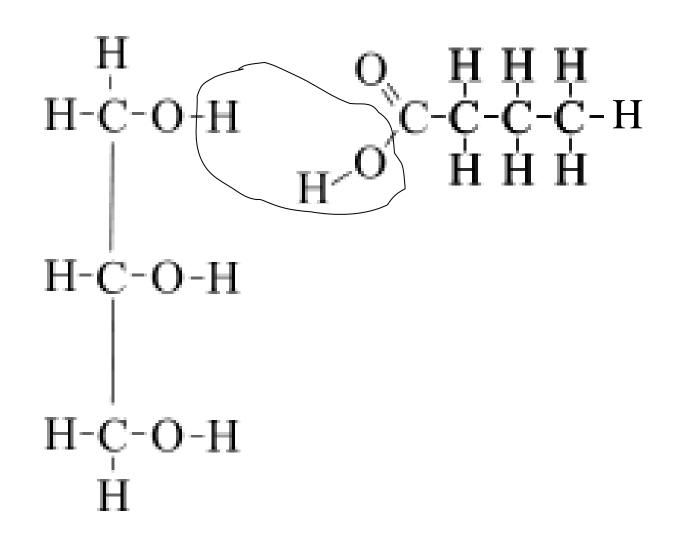
#### 2. Build a molecule of glycerol.

a) Describe the molecule's overall shape.

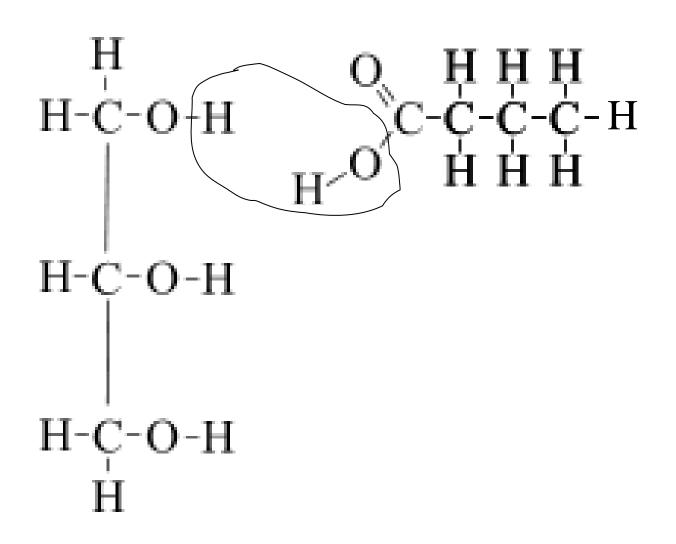




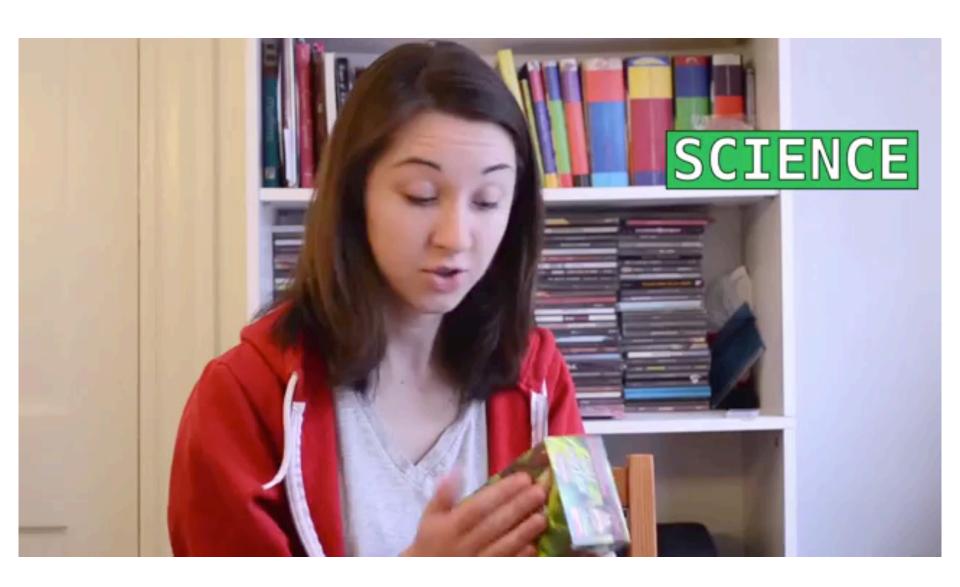
Combine the glycerol and the fatty acids as shown c. Make a schematic diagram of the resulting mole

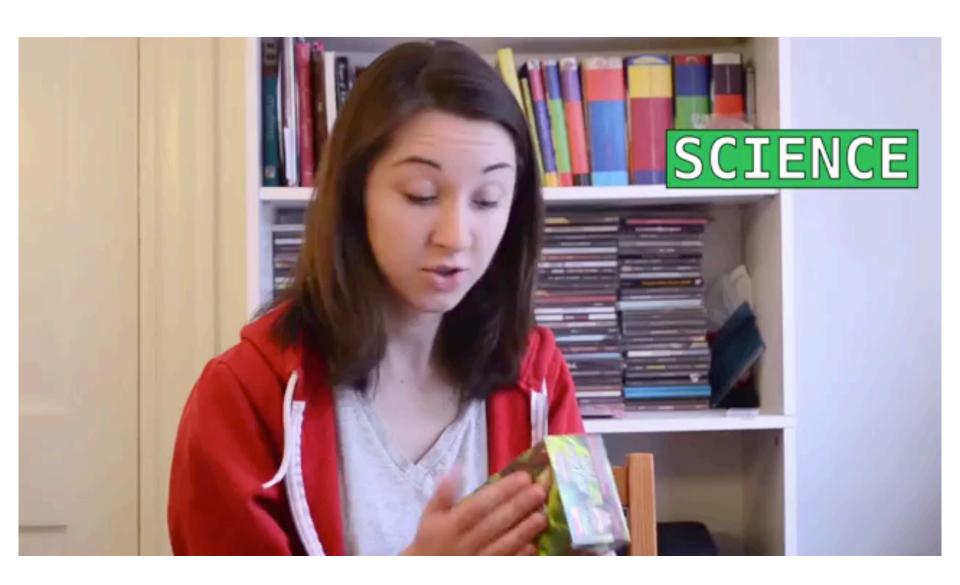


- e. What molecule was created as a byproduct of the synthesis of this fat?
- f. What type of reaction is this?

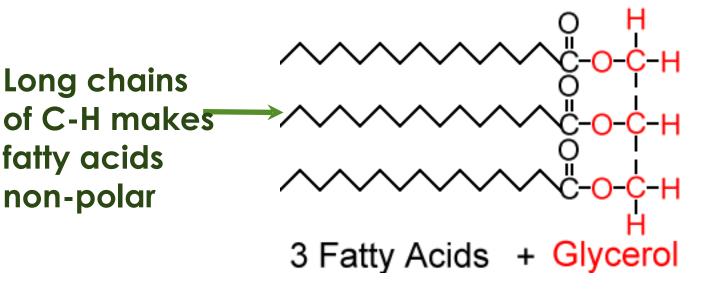


www.BioTopics.co.uk

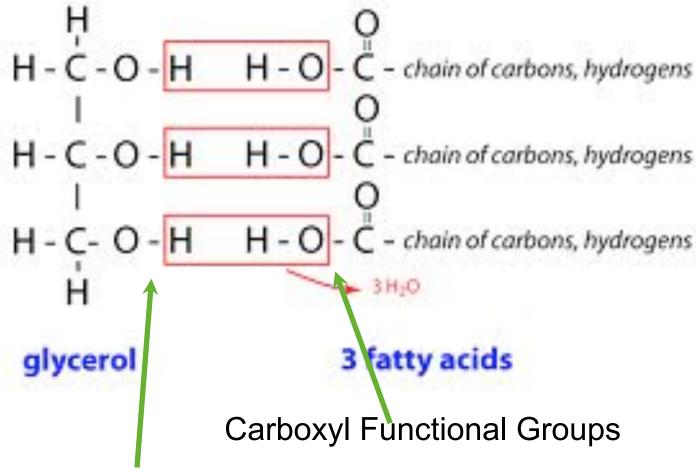




fat molecules have 3 branched structure
 = triglyceride

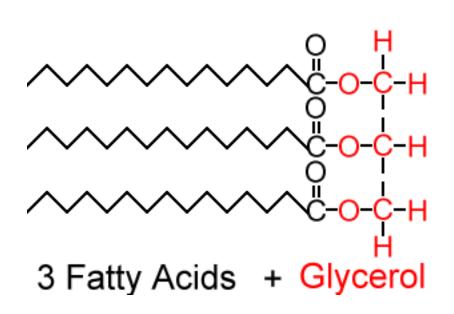


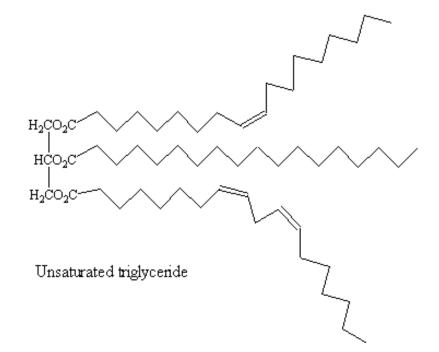
#### Formed by condensation reactions

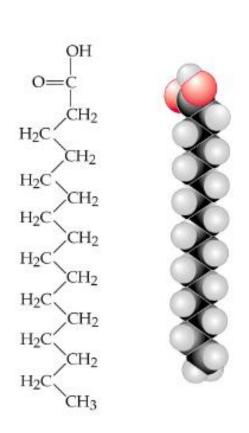


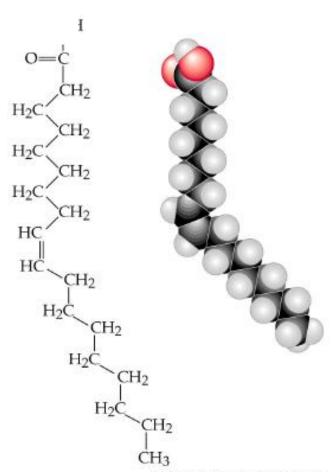
**Hydroxyl Functional Groups** 

 glycerol always has the same structure but fatty acids can be saturated or unsaturated

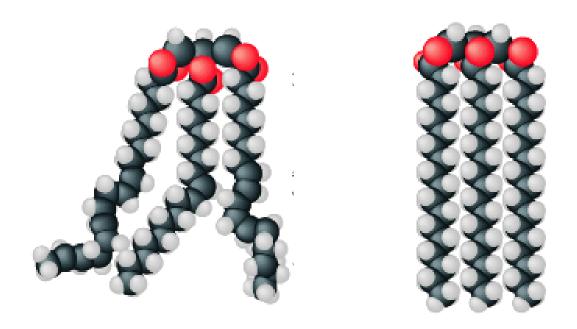






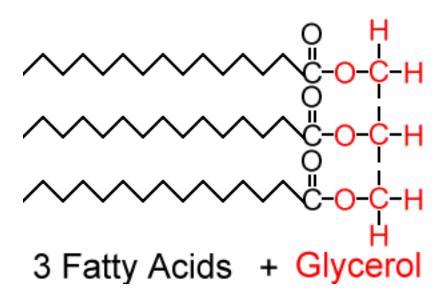


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Saturated: all bonds between carbons are single

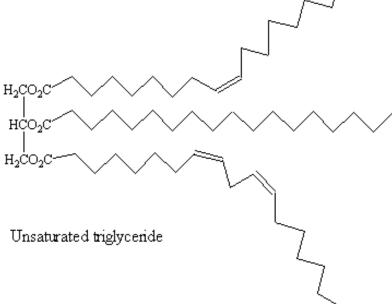
- solid at room temp
- eg. butter



Unsaturated: at least 1 bond is double (note what a double bond does to the shape of the fatty acid)

- liquid at room temp

- eg. Vegetable oil

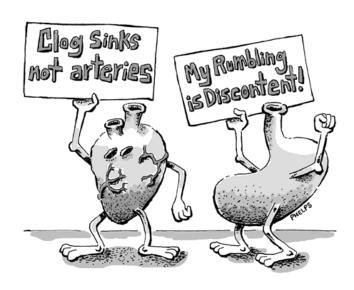


### **Trans Fats**

Saturated Fat

Unsaturated Fat

Trans fat



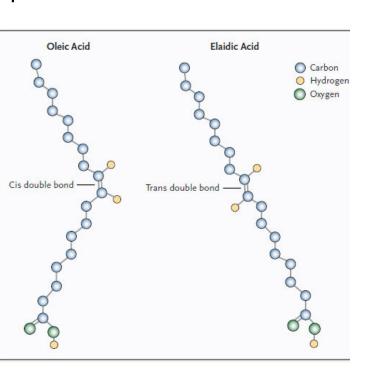
Trans Fat

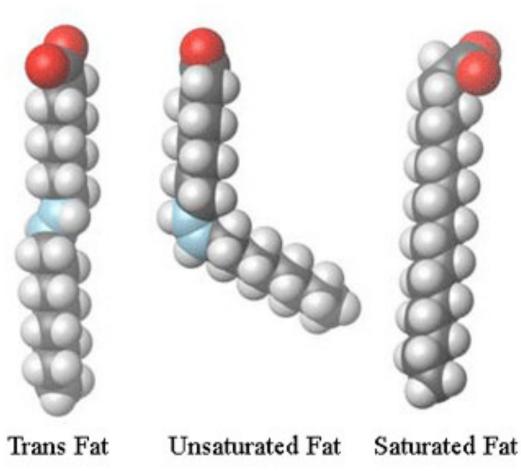




Maasai Diet

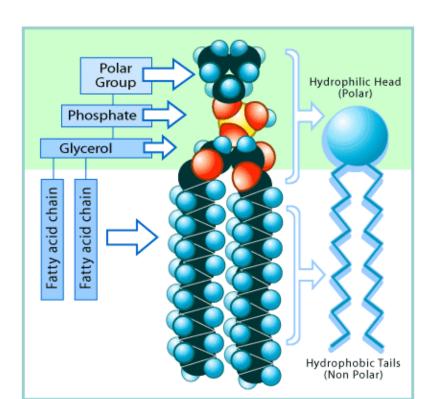
Unsaturated fats can be **Hydrogenated (adding hydrogen)** to make them (i) solid at room temperature & (ii) have a longer shelf life -unfortunately they often become Trans-fats in the process



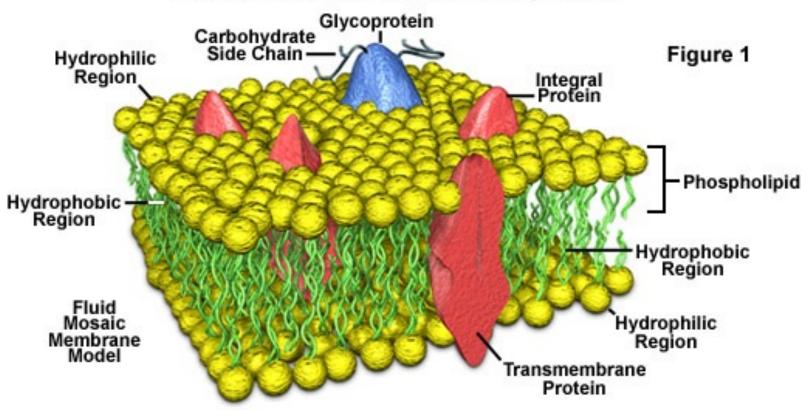


# Phospholipids

- key component of cell membranes
- 1 glycerol + 2 fatty acid chains + 1 phosphate group
- phosphate end is polar and water-soluble, fatty acid end is non-polar



#### Plasma Membrane Structural Components



# Steroids (Sterols)

- carbon-based 4-ring structure
- used to make hormones such as estrogen and testosterone

$$H_3C_{III}$$
 $CH_3$ 
 $H_3C$ 
 $H$ 

## Assignment

- Read health risks of Fat pg 83-85
  - Make concise notes of risks of:
    - Saturated
    - Unsaturated
    - Trans
    - → include evidence (data) that provides supports
- Complete Data Based questions on Page 85/86