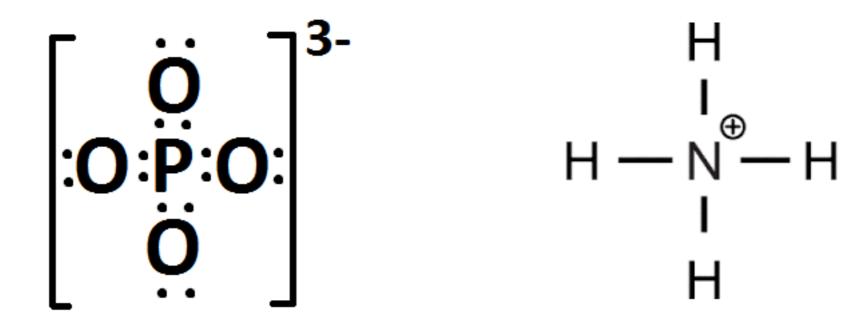
Polyatomic Ions

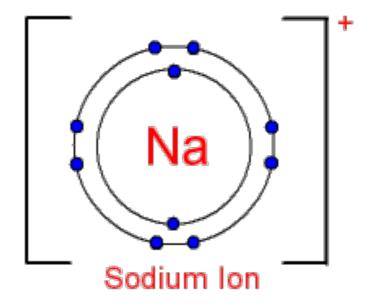


Nitrates in Food



Monatomic Ions

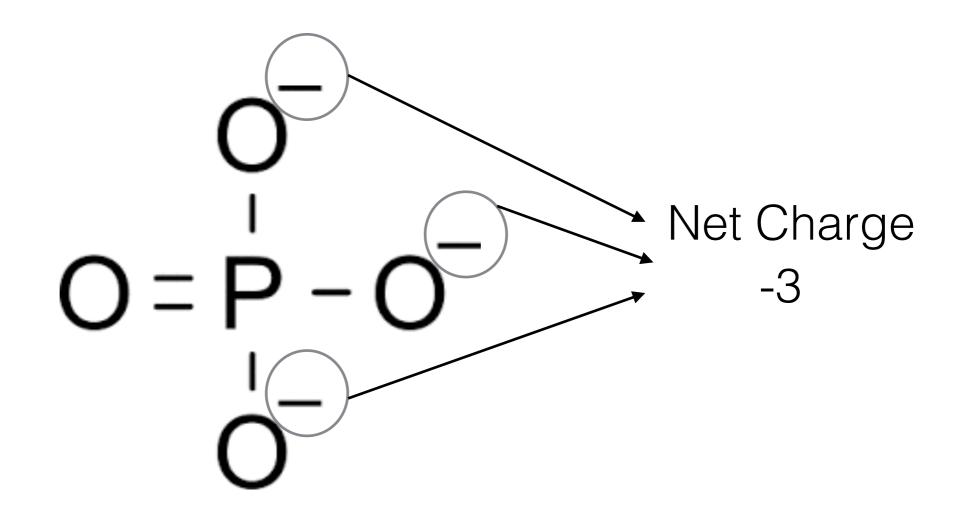
- Ionic compounds we've learned so far consist of monoatomic ions
- Monoatomic lon- an ion composed of only one atom.
- We have only looked at these so far



Polyatomic Ions

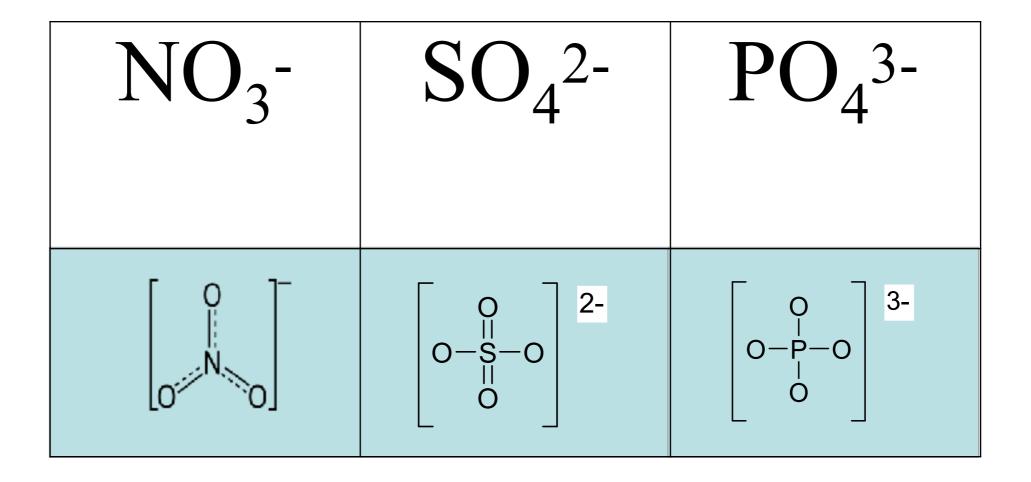
Monoatomic lons- an ion composed of two or more atom types.

Here's one...



Polyatomic Ions

• Example-



Polyatomic Bonding

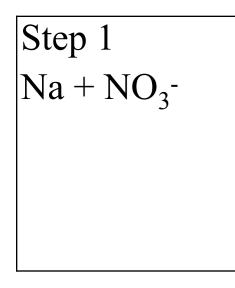
Ionic Bonding with polyatomic ions occurs in the same manner as it does with binary atomic molecules.

a. Balance the charges the same as monoatomic ions

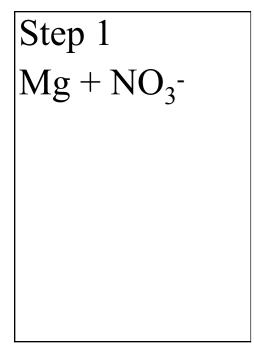
b. Be sure that the charge used is what applies to the whole polyatomic ion.

c. If more than one polyatomic ion is required to balance charges, then use brackets around the polyatomic ion to indicate the number that applies to the whole ion.

Example 1



Example 2



Magnesium Nitrate

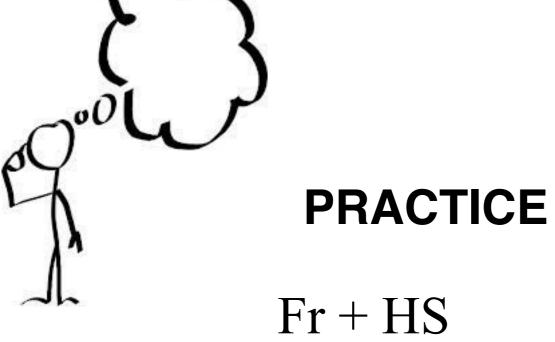


 $Mg + ClO_3$



PRACTICE

 $Na + SiO_3$





$A1 + SO_4$